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Preliminary Communication

Novel secondary allenylidene ruthenium complexes and preparation of *trans*-[(dppm)₂(H)Ru–C≡C–CH(OMe)R] derivatives *

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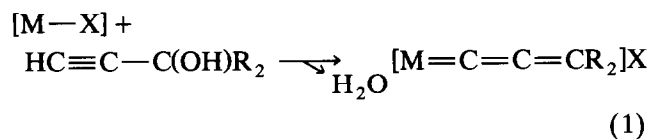
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Abstract

cis-[(dppm)₂Cl₂Ru], **1** (dppm = Ph₂PCH₂PPh₂) reacts with HC≡C–CH(OH)R and NaPF₆ to give *secondary* allenylidene complexes *trans*-[(dppm)₂ClRu=C=C=CHR][PF₆], **3** [R = Ph (**a**), *p*-Ph–Cl (**b**), *p*-PhOMe (**c**), or *trans*-CH=CHPh (**d**)]. Complex **3a** treated with NaBH₄ gives *trans*-[(dppm)₂ClRu–C≡C–CH₂Ph], **6a** and the reaction of **3a–c** with MeONa in MeOH affords *trans*-[(dppm)₂(H)Ru–C≡C–CH(OMe)R], **7a–c**.

1. Introduction

Allenylidene metal complexes have potential as precursors of new carbene derivatives [1] or metal-containing polymers [2,3]. They constitute a possible step toward allenes, *via* transfer of the allenylidene moiety to an organic molecule, and the influence of the allenylidene on the metal reactivity has not yet been shown. The most straightforward route to allenylidene metal complexes consists in the dehydration of propargylic alcohol derivatives promoted by ruthenium(II) complexes (eqn. (1)).



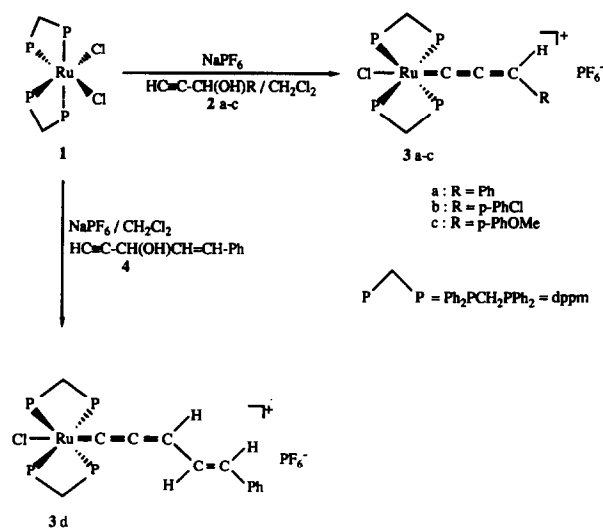
The first example was reported by Selegue [4] with [M–X] sequal to [RuCl(PPh₃)₂C₅H₅], but the allenylidene was stable only for R = Ph. Other examples have recently been reported from [RuCl₂(PR₃)₃(arene)] [5], [N(CH₂CH₂PPh₂)₃RuCl₂] [6], [RuCl₂(dppm)₂] [7]. They only involve *tertiary* propargylic alcohols, for

substitution at carbon C_γ stabilizes the allenylidene metal moiety as does the electron-richness of the metal [5]. However, only one *primary* allenylidene ruthenium intermediate [Ru⁺=C=C=CH₂] has even been postulated [8] and *secondary* allenylidene ruthenium(II) derivative [Ru⁺=C=C=CHR] are likely to be the key intermediate in the polyenylidene ruthenium complex preparation [5] and in the catalytic coupling of propargylic alcohol compounds with allylic alcohols [9].

We now report (i) the formation of the first characterized *secondary* allenylidene metal derivatives *trans*-[(dppm)₂ClRu=C=C=CHR][PF₆] (dppm = Ph₂PCH₂PPh₂) *via* activation of HC≡C–CHOHR with *cis*-[RuCl₂(dppm)₂] and (ii) show that the :C=C=CHR significantly modifies the reactivity of the *trans* Ru–Cl bond, with respect to the :C=C=CR₂, and allows access to new hydridoruthenium complexes.

2. Discussion

As the organometallic moiety (dppm)₂(Cl)Ru⁺ was shown to stabilize the C_γ disubstituted allenylidene complexes towards weak nucleophiles such as alcohols [7], its precursor **1** was allowed to react with HC≡C–CH(OH)R, **2a–c**, in dichloromethane in the presence of NaPF₆ and the red *secondary* allenylidene complexes **3a–c** were isolated in 75–85% yield (Scheme 1).



Scheme 1.

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* Dedicated to Professor M.F. Lappert on the occasion of his 65th birthday.

TABLE 1. Selected NMR data for allenylidene compounds ^a

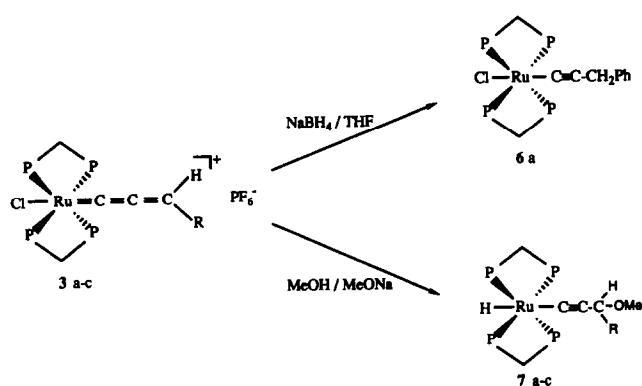
Compound	$\delta(^{31}\text{P})$, ppm PPh ₂	$\delta(^{13}\text{C})$, ppm		
		Ru=C (quint.) (² J _{PC} , ³ J _{CH} , Hz)	Ru=C=C (³ J _{PC} , Hz)	Ru=C=C=C (¹ J _{CH} , Hz) =CH(² J _{PH} , Hz)
3a	-14.91	323.10 (14.4, 6.6)	217.28 (2.5)	150.86 (165.1) 8.49 (2.7)
3b	-15.17	323.24 (14.4, 6.6)	220.24 (2.7)	148.84 (165.5) 8.47 (2.8)
3c	-13.71	308.38 (14.3, 6.3)	200.78 (2.9)	149.92 (163.2) 8.03 (2.6)
3d	-14.59	315.17 (14.3, 6.9)	220.36 (2.7)	149.85 (164.2) 8.06 (2.3)

^a All spectra in CD₂Cl₂ at 297 K.

Analogously, 1-(2-phenylethyl)prop-3-yn-1-ol, **4**, led to the 3-alkenyl allenylidene complex **3d** (77%). The *trans* complexes **3** were characterized by NMR spectroscopy: equivalent ³¹P nuclei, a low field =CHR signal in the ¹H NMR spectrum ($\delta \sim 8.5$ ppm), a quintuplet for the Ru=C signal in the ¹³C NMR spectrum ($\delta \leq 310$ ppm; ²J_{PC}: ~ 14 Hz) and ¹J_{HC γ} = 165 Hz) (Table 1). The absorption among from the C=C=C structure is at $\nu = 1938$ cm⁻¹ (**3a**), much lower than with the *tertiary* complex *trans*-[(dppm)₂ClRu(=C=C=CPh₂)] [PF₆]₅ ($\nu = 1964$ cm⁻¹).

The possibility of nucleophilic addition to the allenylidene at C(α) or C(γ) was studied with hydride. Complex **3a** reacts with sodium borohydride in THF to give the alkynyl derivative **6a** [$\nu(\text{C}\equiv\text{C}) = 2103$ cm⁻¹; 35%], arising from the selective addition of the hydride at the C γ carbon atoms only (Scheme 2, Table 2). The high field signal ($\delta = -5.47$ ppm) of the ³¹P nuclei in NMR spectrum is characteristic of a *trans* Cl-Ru-C \equiv C-R arrangement in related complexes [10].

In contrast, the addition to complex **3** of MeONa in methanol led to an unexpected development. Complexes **7a-c** were obtained and characterized by NMR



Scheme 2.

TABLE 2. Selected NMR data for acetylide compounds ^a

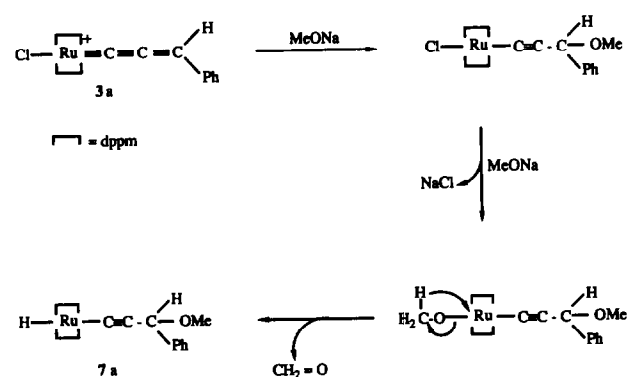
Compound	$\delta(^{31}\text{P})$, ppm PPh ₂	$\delta(^1\text{H})$, ppm H-Ru (² J _{PH} , Hz)	$\delta(^{13}\text{C})$, ppm	
			Ru-C \equiv C (² J _{PC} , Hz)	Ru-C \equiv C
6a	-5.47	-	101.44 (15.6)	105.94
7a	3.08	-7.19 (20)	123.64 (13.4)	109.89
7b	3.04	-7.16 (19.9)	124.96 (13.4)	109.23
7c	3.00	-7.21 (19.9)	123.12 (13.4)	110.14

^a All spectra in CD₂Cl₂ at 297 K.

spectroscopy (Table 2). Their spectra are consistent with the addition of the methoxide to the C γ carbon atom and the conversion of the *trans* Ru-Cl bond into a Ru-H bond. The reaction of **3a** with CD₃ONa in CD₃OD led to the sole formation of *trans*-[(dppm)₂(D)Ru-C \equiv C-CH(OCD₃)Ph], **8** isolated in 30% yield [$\delta(\text{ppm}) = 3.15$ (PPh₂), = 4.75(s, =C-CH(OCD₃), correct elemental analysis). The formation of the Ru-H bond likely takes place after the nucleophilic addition of MeO⁻ at C γ and results from substitution of the chloride by the methoxide, followed by formaldehyde elimination (Scheme 3).

This formation of an Ru-H bond *trans* to the allenylidene would not be observed under similar conditions with C γ disubstituted complexes such as *trans*-[(dppm)₂ClRu=C=C=CPh₂][PF₆]₅, for which addition of the methoxide at C γ only took place, giving *trans*-[(dppm)₂ClRu-C \equiv C-C(OMe)Ph₂]**9**. It is likely that, in the intermediates *trans*-[(dppm)₂ClRu-C \equiv C-C(OMe)(R)Ph] the C \equiv C-CH(OMe)Ph group favours *trans*-chloride substitution.

The isolation of complex **3** show that non-heteroatom-stabilized *secondary* allenylidene ruthenium complexes are stable, and their transformations suggest that *secondary* allenylidene ligands are able to introduce subtle modifications in the reactivity of ruthenium complexes.



Scheme 3.

Acknowledgment

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